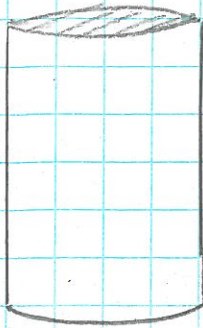


water



Volume, V_w

Mass, m_w

density $\rho_w = 1000 \frac{\text{kg}}{\text{m}^3}$

Now,

$$m_w = m_g$$

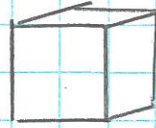
$$\rho_w V_w = \rho_g V_g$$

So:

$$V_w = V_g \left(\frac{\rho_g}{\rho_w} \right) = \underline{1930 \text{ cm}^3}$$

Gold

$\frac{18-1}{1}$



Volume $V_g = 100 \text{ cm}^3$

Mass, m_g

density, $\rho_g = 19,300 \frac{\text{kg}}{\text{m}^3}$